

Synthesis of the Trichothecene Mycotoxin, T-2 Tetraol

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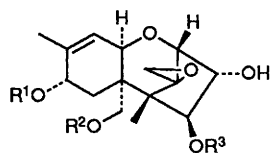
The synthesis, as its tetra-acetate (neosolaniol diacetate) (**21**), of T-2 tetraol (**1**), the parent member of a group of trichothecene mycotoxins including the highly toxic T-2 toxin (**2**) and HT-2 toxin (**3**), is described.

The trichothecene mycotoxins constitute a major group of complex fungal sesquiterpenoids.¹ They possess a wide range of largely adverse biological behaviour, including antibacterial, antiviral, cytostatic and phytotoxic activity. Despite some 80 non-macrocyclic trichothecenes being known, only a few have yielded to total synthesis;² these include trichodermin,³ trichodermol,⁴ verrucarol,⁵ anguidine,⁶ calonectrin,⁷ and 12,13-epoxytrichothec-9-ene.⁸

Continuing our studies on the syntheses^{3,9} and synthetic transformations¹⁰ of the trichothecene mycotoxins, we communicate the first synthesis of the complex trichothecene, T-2 tetraol (**1**). The synthetic sequence is outlined in Scheme 1. Our synthetic strategy consisted of five key elements: (a) construction of the *cis*-fused *AB* ring system by a Diels–Alder cycloaddition, (b) formation of ring *C* by an intramolecular

aldol reaction, (c) creation of the 3 α ,4 β diol functionality of ring *C* by a stereoselective α -oxygenation/reduction protocol, (d) ring *A* enone formation *via* a regio-specific thermodynamically controlled α -selenylation, and (e) regio- and stereospecific reduction of this enone to the required α -alcohol.

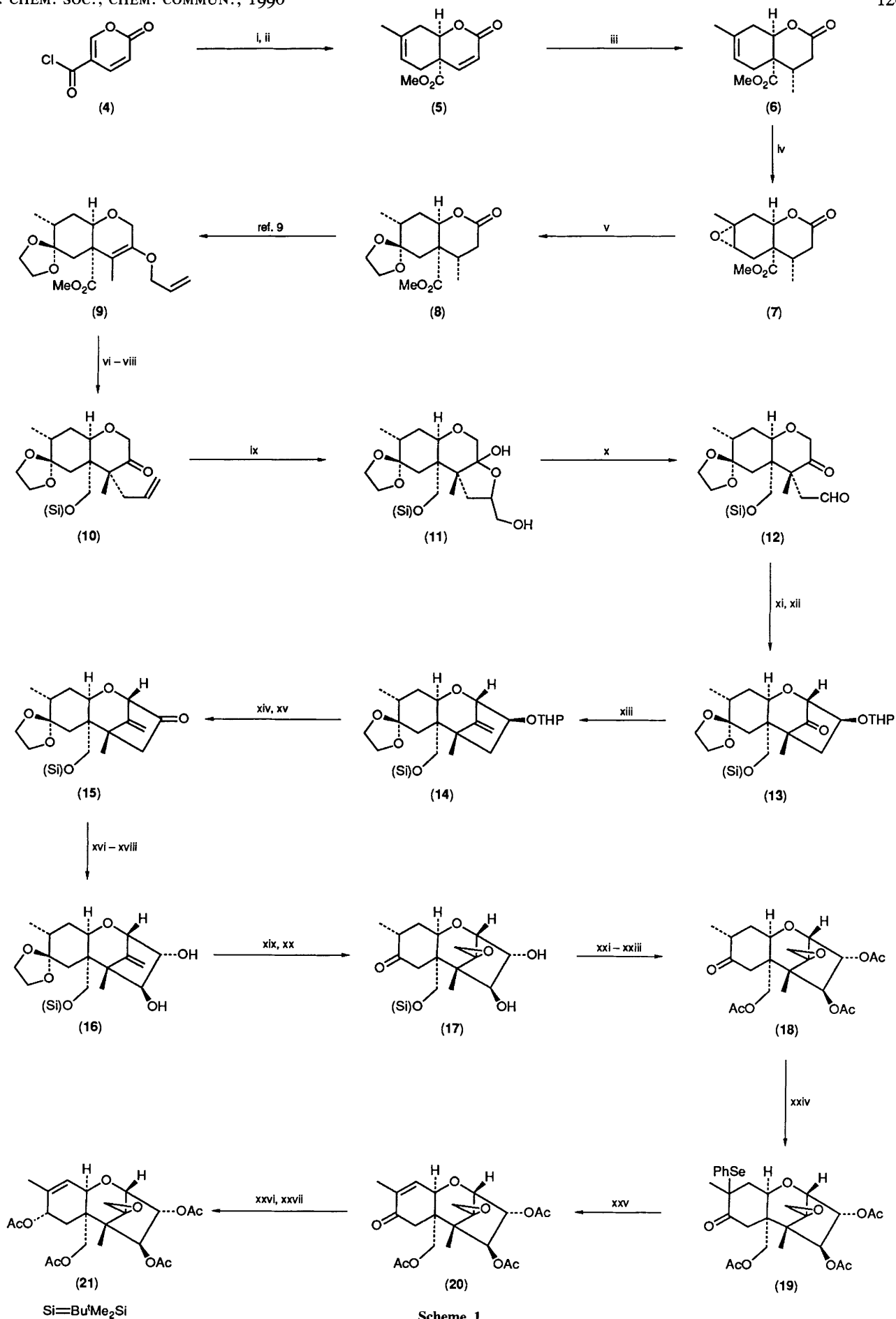
Scheme 1. Reagents and conditions: i, isoprene, toluene, 120 °C (sealed tube), 1 h; ii, MeOH, Et₃N, 0 °C, 41% overall; iii, LiCuMe₂, Me₃SiCl, Et₂O, –78 °C to room temp., 55%; iv, *m*-chloroperbenzoic acid (MCPBA), Na₂HPO₄, CH₂Cl₂, 0 °C, 2.5 h, 81%; v, ethane-1,2-diol, BF₃·Et₂O, CH₂Cl₂, room temp., 18 h, 61%; vi, LiAlH₄, Et₂O, 0 °C, 0.5 h, 94%; vii, toluene, Et₃N (5%), 110 °C, 24 h, 89%; viii, Bu^tMe₂SiOSO₂CF₃, 2,6-lutidine, CH₂Cl₂, room temp., 0.5 h, 89%; ix, NaIO₄, OsO₄ (cat.), Et₂O, H₂O, room temp., 18 h, 88%; x, H₃IO₆, Et₂O, room temp., 5 min; xi, NaOMe, MeOH, reflux, 0.5 h, 62% overall; xii, dihydropyran, PPTS, room temp., 12 h; xiii, Ph₃P=CH₂, tetrahydrofuran (THF), hexane, reflux, 18 h; xiv, ethane-1,2-diol, PPTS, CH₂Cl₂, reflux, 18 h, 61% overall; xv, CrO₃, 3,5-dimethylpyrazole, CH₂Cl₂, room temp., 1 h, 81%; xvi, KN(SiMe₃)₂, toluene, THF, 0 °C, 1 h; xvii, 2-*p*-tolylsulphonyl-3-(*p*-nitrophenyl)oxaziridine, THF, –78 °C to room temp.; xviii, NaBH₄, MeOH, 0 °C, 0.5 h, 37% overall; xix, MeOH, PPTS, room temp., 48 h, 90%; xx, MCPBA, Na₂HPO₄, CH₂Cl₂, room temp., 48 h, 50%; xxi, Ac₂O, pyridine (py), Et₂O; xxii, Bu₄NF·3H₂O, THF, room temp., 48 h; xxiii, Ac₂O, py, 84% overall; xxiv, PhSeCl, PPTS, AcOEt, room temp., 48 h, 85%; xxv, O₃, CH₂Cl₂, –78 °C to room temp., 72%; xxvi, LiBu^s₃BH, THF, –78 °C⁵ to room temp., 1 h; xxvii, Ac₂O, py, 88% overall.



(1) R¹ = R² = R³ = H

(2) R¹ = (Me)₂CHCH₂CO, R² = R³ = Ac

(3) R¹ = (Me)₂CHCH₂CO, R² = Ac, R³ = H



Scheme 1

Cycloaddition¹¹ of coumaly chloride (**4**) with isoprene followed by esterification of the product mixture gave the adduct (**5**)[†] and its regioisomer (isomeric ratio 4:1). Treatment of this mixture with lithium dimethyl cuprate gave, after purification,[‡] the lactone (**6**) as a single methyl epimer. Epoxidation provided the epoxide (**7**), again as a single epimer, reaction of which with ethane-1,2-diol in the presence of $\text{BF}_3 \cdot \text{Et}_2\text{O}$ provided the ketal (**8**).

Following established methodology,⁹ two-step oxidation of the lactone enolate to the α -ketolactone, followed by allylation of this in its enol form and reductive deoxygenation gave the allyl enol ether (**9**). Hydride reduction of the ester function, followed by Claisen rearrangement and alcohol protection using *t*-butyldimethylsilyl trifluoromethanesulphonate¹² provided the α -allyl ketone (**10**) and its epimer (epimeric ratio 3.2:1). Attempted ozonolysis, to generate the required aldehyde moiety, proved unsuccessful. Remarkably, catalytic osmylation in the presence of sodium metaperiodate stopped at the diol stage, with isolation of the tetrahydrofuran (**11**). However, treatment of this with ethereal periodic acid¹³ resulted in clean cleavage to the desired aldehyde (**12**).

Aldol cyclisation followed by pyridinium toluene-*p*-sulphonate (PPTS) catalysed¹⁴ tetrahydropyranylation gave the full tricyclic system, as a mixture of 3β (**13**) and 3α epimers (epimeric ratio 4.2:1). The more abundant 3β epimer was subjected to the Wittig reaction to give the alkene (**14**). Selective tetrahydropyran (THP) ether deprotection, using ethane-1,2-diol in the presence of PPTS, followed by chromium trioxide-3,5-dimethylpyrazole¹⁵ oxidation led to the ketone (**15**).

Oxaziridine-mediated α -hydroxylation¹⁶ of the enolate of this ketone, followed by hydride reduction gave the desired $3\alpha,4\beta$ diol (**16**), both reagents having attacked from the less hindered *exo* face of the oxabicyclo[3.2.1]octane sub-unit.¹⁷ Selective acetal deprotection followed by epoxidation gave the keto-epoxide (**17**).

Fluoride-induced deprotection and per-acetylation gave the keto-triacetate (**18**). This ketone was identical in all respects, apart from being racemic, with the ketone obtained by catalytic hydrogenation of the naturally derived^{10b} enone (**20**). This latter homochiral ketone (**18**) was used to complete the synthesis. PPTS-catalysed α -selenylation provided the selenide (**19**) as a mixture of epimers, but regioisomerically pure. Selenoxide formation and cycloelimination then provided the

enone (**20**). Regio- and stereo-specific reduction of this enone using lithium tri-*S*-butylborohydride provided the 8α alcohol. Acetylation provided T-2 tetraol tetra-acetate (**21**), $[\alpha]_{\text{D}}^{25} +55.5^\circ$ (*c* 0.128 in AcOEt), m.p. 182–183 °C (lit.,¹⁸ 178–179 °C), with spectral properties identical in all respects with literature data.^{18,19}

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[†] Compounds (**5**)–(**18**) are racemic; all other stereochemically defined compounds are homochiral.

[‡] All reported compounds were fully characterised by elemental analysis and/or high resolution mass spectrometry, and IR and ¹H and ¹³C NMR spectroscopy.